

A REVIEW

**ORCHID NUTRITION
& FERTILISERS**

March 2024

Peter Young

HOW DO NATIVE SPECIES GET NUTRIENTS?

- ❖ Thunder storms – lightning with rain → Nitrogen
- ❖ Sweat from trees – ‘Guttation’ → soil minerals
- ❖ Symbiotic fungi – Mycorrhiza in root coating (Velamen)
- ❖ Decomposing plant material → organic Nitrogen
- ❖ Rock weathering releasing minerals - lithophytes
- ❖ Gaseous absorption – mostly cloud forests → Nitrogen
- ❖ Others – monkeys, bird nests etc

**Thunderstorm + Lightning combines Nitrogen Gas +
Oxygen → Nitrogen Oxide**



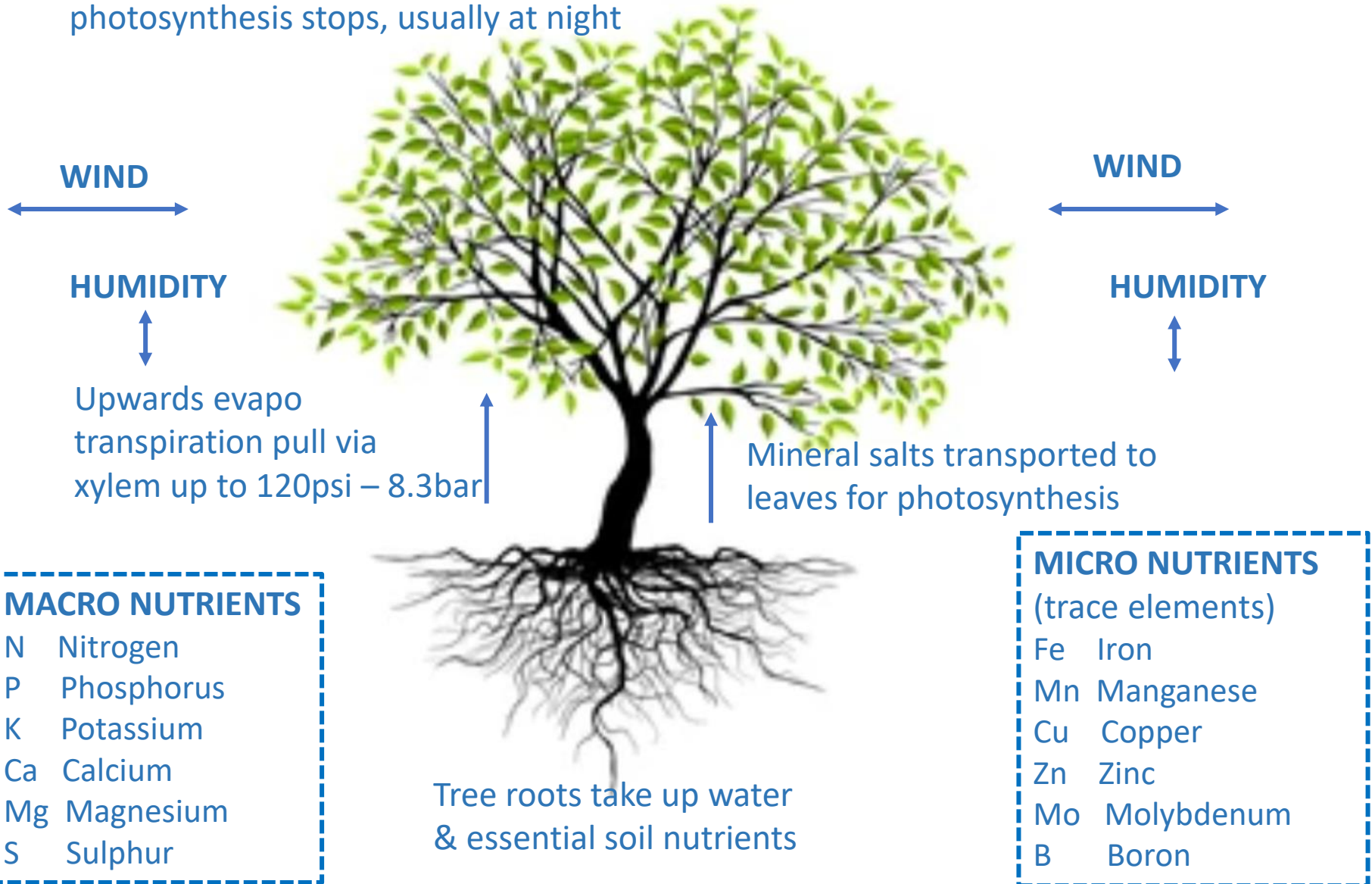
Nitrogen Oxide + Rain (H_2O) → Nitrate (NO_3)

**The World's agriculture, grasslands and forests rely
on natural Nitrogen fertiliser**

PLANT GUTTATION:

The action of plant leaves sweating out mineral salts in sap droplets via specialist porous openings called Hydathodes.

This allows trees to remove excess water as required and occurs mostly when photosynthesis stops, usually at night



 mineral content of plant g...

What does the water of guttation contain?



Guttation is the loss of water in solution form through hydathodes. The water lost in guttation is **a dilute solution of mineral salts**.

 <https://www.toppr.com> › question

[Water of guttation is - Toppr](#)

[MORE RESULTS](#)

What is the primary substance excreted from the plant via guttation?



The water carrying dissolved minerals that travels upward through the xylem vessels is called the **xylem sap**. Therefore, the primary substance excreted from a plant via guttation is the xylem sap.

 <https://www.nagwa.com> › explainers

[Lesson Explainer: Excretion in Plants - Nagwa](#)

[MORE RESULTS](#)

What does guttation in plants result



Discover



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Guttation - Wikipedia

Visit >

WHAT CAN WE LEARN FROM NATURE?

- ❖ Little but often, in particular Nitrogen
- ❖ Importance of air movement & ventilation
- ❖ Roots dry out between rainfall/fertilising events
- ❖ Species growth & flowering linked to weather patterns
- ❖ Species only found where all factors line up
- ❖ Orchids can re-hydrate during high humidity conditions
- ❖ Some need light/wind/exposure/dry season
- ❖ Others need shade, wind protection/regular moisture

‘MAJOR’ ELEMENTS & ROLE

Demand influenced by light & exposure (air movement)

N – Nitrogen

- Determines demand of all other nutrients
- NO_3 Nitrate via leaves; NH_4 Ammonium via roots

P – Phosphorus

- Strong roots, stem extension, flower colour, size and numbers

K - Potassium

- Strength of whole plant including roots, flower brightness & quality, competes with Calcium & Boron uptake

Ca – Calcium

- Disease prevention, leaf & flower shape & durability, stem extension, tolerance to weather extremes

Mg – Magnesium

- Sunscreen in plant & maintains high chlorophyll, competes against Calcium & Potassium

S – Sulphur

- Maintains high chlorophyll and helps plant use Nitrogen & P, aids cold hardiness

‘TRACE ELEMENTS’ & ROLE

Demand influenced by ‘major’ nutrient levels

Fe – Iron

- Important to maintain chlorophyll in emerging new growth

Mn – Manganese

- Disrupted by over use of Lime/Dolomite etc. or if water & media is +pH7

Cu – Copper

- Important for maintaining upright growth & strong roots

Zn – Zinc

- Maintains strong growth & full size leaves

Mo – Molybdenum

- Helps Nitrogen uptake in cold weather, strong pollen

B – Boron

- Important in Calcium uptake & movement – critical for pollination & seed development



calcium content of gypsum

All

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Per ton

Calculator

Formula

In kg

Pdf

Pure gypsum contains **23.3%** calcium (Ca) and 18.6% sulfur (S). Gypsum is moderately soluble in water (2.5 g per L) or approximately 200 times greater than lime (CaCO_3). This makes the calcium in gypsum more mobile than the calcium in lime and allows it to more easily move through the soil profile.

Compound	Formula	Solubility (kg/100L) @ 20° C
Calcium Carbonate (Lime)	CaCO_3	0.0013
Calcium Oxide (Burnt or Quick Lime)	CaO	0.13 (at 25 deg C)
Calcium Hydroxide (Slaked or Hydrated Lime)	Ca(OH)_2	0.17
Calcium Sulfate (Gypsum)	$\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$	0.26
Monocalcium Phosphate (Superphosphate)	$\text{Ca(H}_2\text{PO}_4)_2 \cdot \text{H}_2\text{O}$	1.8
Calcium Chloride	$\text{CaCl}_2 \cdot 6\text{H}_2\text{O}$	75
Calcium Nitrate	$\text{Ca(NO}_3)_2 \cdot 4\text{H}_2\text{O}$	129

NITRATE NITROGEN (NO_3) & ORCHIDS

- Raises media pH
- Mostly absorbed by leaves
- Absorbed by root tips only
- Comes from thunderstorms
- Guttation from trees sweating
- Nitrifying bacteria in potting mix
- Fertiliser – check label

AMMONIUM NITROGEN (NH_4) & ORCHIDS

- **Lowers media pH**
- **Mostly absorbed by all roots**
- **None in thunderstorm rain**
- **Guttation from trees sweating**
- **Decomposing organic matter – organic**
- **Orchid mycorrhiza in roots**
- **Urea in fertiliser – check label**



(PDF) Phalaenopsis ...
researchgate.net



ResearchGate



Join for free

Login

Article

PDF Available

Phalaenopsis can absorb urea directly through their roots

June 2008 · Plant and Soil 319(1):95-100

DOI:[10.1007/s11104-008-9852-5](https://doi.org/10.1007/s11104-008-9852-5)

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Agrarian Sciences • An. Acad. Bras. Ciênc. 90
(04) • Oct-Dec 2018 •

<https://doi.org/10.1590/0001-3765201820171008>

 COPY

Impact of Nitrate and Ammonium ratio on Nutrition and Growth of two Epiphytic Orchids



AUTHORSHIP

SCIMAGO INSTITUTIONS RANKINGS

Abstract

Phalaenopsis and *Dendrobium* do not grow and flower well with 100% ammonium ($\text{NH}_4\text{-N}$); and there are detailed studies on the effects of nitrate ($\text{NO}_3\text{-N}$) and ammonium ratios on the flowering, but no information about accumulation of other nutrients

50% Nitrate & 50% Ammonium Benefits growth



ScienceDirect



View PDF

[Download full issue](#)

Plant Diversity

Volume 40, Issue 4, August 2018, Pages 196-208

Physiological diversity of orchids

Shibao Zhang^{a 1}  , Yingjie Yang^{a b 1},
Jiawei Li^{a b}, Jiao Qin^a, Wei Zhang^a, Wei Huang^a,
Hong Hu^a

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<https://doi.org/10.1016/j.pld.2018.06.003> 

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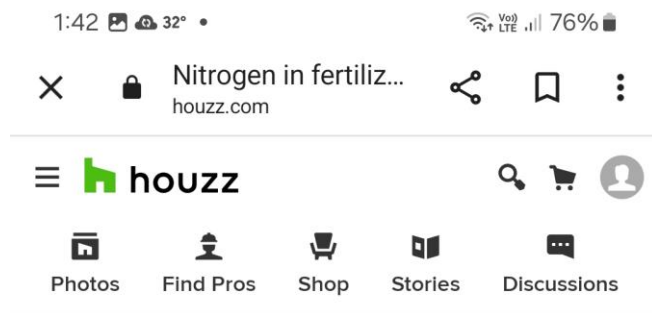
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Abstract

FEEDBACK 





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Orchid

Nitrogen in fertilizers and the truth about Urea



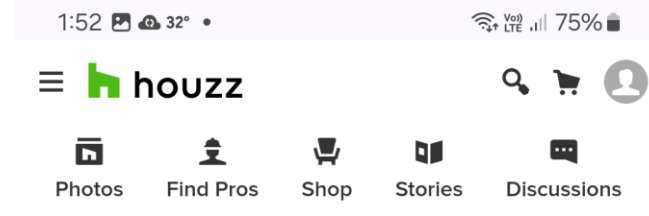
painterart

15 years ago

Sorry for being late on the discussion about fertilizer. It is all a matter of finding time. the one thing I want to dispel is that urea is not good for plants or orchids contrary to often what is written about it. I posted a similar post to this one years ago on this forum, but information gets lost especially when there is a lot of misinformation.

Almost all published studies done on fertilizers and plant nutrients have been done on crop plants. I do think though that much of this is transferable to orchids.

Most orchids can absorb nitrogen directly in the forms of cationic ammonium ion (NH_4^+), the anion nitrate (NO_3^-), and urea. The nitrogen in the ammonium ion (NH_4^+) is available immediately to the plant for the production of amino acids and



plants. I do think though that much of this is transferable to orchids.

Most orchids can absorb nitrogen directly in the forms of cationic ammonium ion (NH_4^+), the anion nitrate (NO_3^-), and urea. The nitrogen in the ammonium ion (NH_4^+) is available immediately to the plant for the production of amino acids and other compounds. Nitrate nitrogen (NO_3^-) on the other hand has to be reduced in order to be used by plants.

Nitrate nitrogen is more "time released" than urea. Urea's nitrogen availability is more the middle ground between nitrate and ammonium. Urea hydrolyses into ammonium (NH_4^+) and CO_2 . This breakdown of urea can take place in the surrounding soil, bark or coir and in the roots and leaves of orchids. The urease enzyme breaks down the urea in leaves and roots. Urea is readily absorbed by roots and leaves and can be used in foliar feeding. While urea is broken down in a growing medium and the resulting ammonium nitrogen is absorbed by the orchid, it is not necessary as orchid can directly absorb urea and break it down within the orchid. Contrary to older orchid book statements urea does not take a year to break down, and can be used by orchids directly.

New Phytologist / Volume 213, Issue 1 / p. 10-12

Commentary | [Free Access](#)

Nitrogen transport in the orchid mycorrhizal symbiosis – further evidence for a mutualistic association

John D. W. Dearnaley  Duncan D. Cameron

First published: 28 November 2016

<https://doi.org/10.1111/nph.14357>

Citations: 46

This article is a Commentary on Fochi *et al.*, **213**: 365–379.

About | Sections



Abstract

This article is a Commentary on Fochi *et al.*, **213**: 365–379.

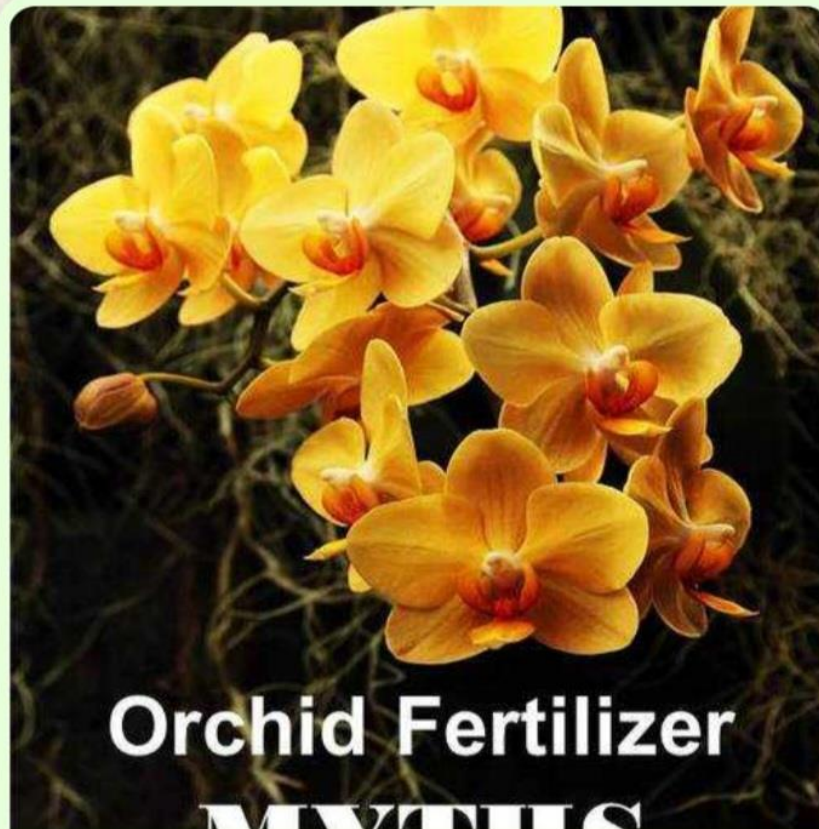
Introduction

Mycorrhizal symbiosis is integral to the

GROWER ATTITUDES TOWARDS NUTRIENTS & FERTILISERS

OPINIONS & ATTITUDES ALL OVER THE PLACE AS FOLLOWS:

- **Doesn't matter what fertiliser you use?**
- **The plant only takes what it needs?**
- **Rest goes down the drain?**
- **Plenty of Calcium & Magnesium in water?**
- **Water pH doesn't really matter?**
- **Don't use Urea based fertiliser?**
- **Only use Nitrates – Nitrate vs Urea?**



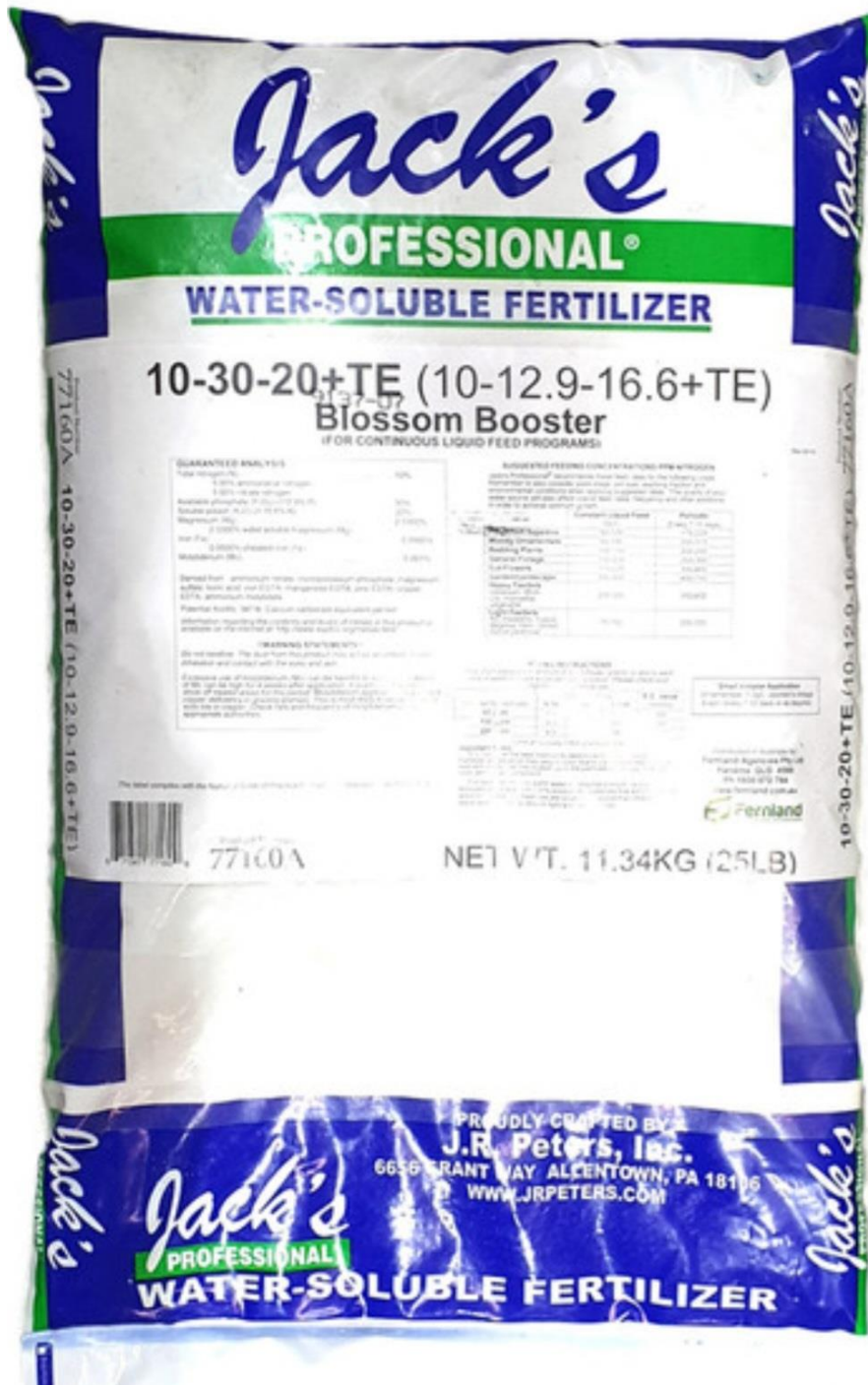
Orchid Fertilizer

Orchid Fertilizer Myths - Grow More Flowers

There is a lot of information floating aro...
www.gardenmyths.com

https://www.gardenmyths.com/orchid-fertilizer-myths/#There_is_No_Such_Thing_as_Orchid_Fertilizer

10:16 am ✓✓



EQUAL NITRATE &
AMMONIUM

HIGH P
NO Ca

USA

N-P₂O₅-K₂O

AUSTRALIA

(N-P-K)

VoLTE 76%

10-30-20+TE (10-12.9-16.6+TE)

Blossom Booster

(FOR CONTINUOUS LIQUID FEED PROGRAMS)

Rev 05/19

Product Number

77160A

0+TE (10-12.9-16.6+TE)

GUARANTEED ANALYSIS

Total nitrogen (N) 10%
 5.00% ammoniacal nitrogen
 5.00% nitrate nitrogen
 Available phosphate (P₂O₅) (=12.9% P) 30%
 Soluble potash (K₂O) (= 16.6% K) 20%
 Magnesium (Mg), 2.0300%
 2.0300% water soluble magnesium (Mg)
 Iron (Fe) 0.0500%
 0.0500% chelated iron (Fe)
 Molybdenum (Mo) 0.001%

Derived from: ammonium nitrate, monopotassium phosphate, magnesium sulfate, boric acid, iron EDTA, manganese EDTA, zinc EDTA, copper EDTA, ammonium molybdate

Potential Acidity: 347 lb. Calcium carbonate equivalent per ton.

Information regarding the contents and levels of metals in this product is available on the internet at: <http://www.aapco.org/metals.html>

!!WARNING STATEMENTS!!

Do not swallow. The dust from this product may act as an irritant. Avoid inhalation and contact with the eyes and skin.

Excessive use of molybdenum (Mo) can be harmful to stock. Plant levels of Mo can be high for 4 weeks after application. It is advisable to keep stock off treated areas for this period. Molybdenum application may induce copper deficiency in grazing animals. This is most likely to occur on sandy soils low in copper. Check rate and frequency of molybdenum use with appropriate authorities.

SUGGESTED FEEDING CONCENTRATIONS PPM NITROGEN

Jack's Professional® recommends these feed rates for the following crops. Remember to also consider plant stage, pot size, leaching fraction and environmental conditions when applying suggested rates. The quality of your water source will also affect overall feed rates, frequency and other additions in order to achieve optimum growth.

	Constant Liquid Feed (CLF)	Periodic (Every 7-10 days)
Plugs/Salt Sensitive	50-125	175-225
Woody Ornamentals	50-100	200-375
Bedding Plants	100-150	200-250
General Foliage	100-200	250-300
Cut Flowers	175-225	300-450
Garden/Landscape	200-300	400-750
Heavy Feeders Geranium, Mum, Lily, Poinsettia, Vegetable	200-300	350-400
Light Feeders NG impatiens, Fuschia, Begonia, Fern, Orchid, Native perennial	75-150	200-250

MIXING INSTRUCTIONS

This chart displays the amount of dry fertilizer (grams) to add to each litre of water to make a concentrated solution. Please check your injector setting before use.

Desired N feed rate	Injector Setting			E.C. value (mmhos)
	1:15	1:100	1:200	
50 ppm	1.0	6.75	13.5	.49
100 ppm	2.0	13.5	27.0	.98
200 ppm	4.0	27.0	54.0	1.96

Limit of Solubility = 600 grams per litre

Important Notes:

E.C. value is the best method to determine fertilizer strength. Fertilizer appearance may vary in color due to variation in raw material

Small Volume Application

Ornamentals: 1-2g/L (applied to foliage & soil) every 7-10 days or as required.

Distributed in Australia by:
Fernland Agencies Pty Ltd



AMMONIUM
(no Nitrate)
+
SULPHATES

WITH
Ca & Mg

SECTION 3 COMPOSITION / INFORMATION ON INGREDIENTS

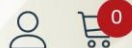
Substances

See section below for composition of Mixtures

Mixtures

CAS No	%[weight]	Name
Not Available	>40	chicken manure
Not Available	<5	seaweed
7783-20-2	<20	<u>ammonium sulfate</u>
7778-80-5	<25	<u>potassium sulfate</u>
7783-28-0	<11	<u>diammonium phosphate</u>
Not Available	<3	rock phosphate
7758-98-7	<1	<u>copper sulfate</u>
10101-68-5	<1	<u>manganese sulfate, hydrate</u>
7487-88-9	<3	<u>magnesium sulfate, anhydrous</u>
7720-78-7	<2	<u>ferrous sulfate anhydrous</u>
7631-95-0	<1	<u>sodium molybdate</u>
1303-96-4	<1	<u>sodium borate, decahydrate</u>
7733-02-0	<1	<u>zinc sulfate</u>

SECTION 4 FIRST AID MEASURES

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ICL Specialty Fertilisers

Peters Excel CalMag Finisher



No Reviews Yet

[Write A Review](#)

Trade/Business Customers:

NITRATE ONLY

HIGH Ca, Mg & K



Peters
Excel

CalMag Finisher

13-2.2-16.6+5Ca+1.2Mg+TE



Peters Excel CalMag Finisher is often used as a follow-up fertilizer to Peters Excel CalMag Grower 15-5-15. Its high potassium formula leads to a more compact growth. This Peters Excel product is a 'single-tank mix' solution providing plants with all the necessary nutrients from a single tank.

Advantages

- Especially developed for use in soft water, unique in its effects.
- Healthy growth as a result of continuous supply of calcium and magnesium.
- Reliable system thanks to selected composition of raw materials and chelated trace elements.
- High level of chelated trace elements ensures perfect color and growth.
- Maximum availability and absorbability because of M-77 chelating complex.
- Single-tank-mix concept: all nutrient elements in a single storage tank, including calcium.

Directions for use

The product can be mixed with calcium nitrate in the same tank (50:50%).

To ensure that this product dissolves completely, prepare the stock solution 1-2 hours before use, stir well or use warm water. Please contact your ICL Specialty Fertilizers advisor for a specific crop recommendation. Do not mix Peters Excel with any other NPK or phosphate containing fertilizer, except phosphoric acid.

Store under dry conditions. Partly used or damaged bags should be closed well.

elemental	
Nitrogen Total (N)	13%
Nitrate nitrogen (NO ₃ -N)	11.0%
Urea nitrogen (Ur-N)	2.0%
Phosphorus (P)	2.2%
Water soluble	2.2%
Potassium (K)	16.6%
Water soluble	16.6%
Magnesium (Mg)	1.2%
Water soluble	1.2%
Calcium (Ca)	5.0%
Water soluble	5.0%
Iron (Fe)	0.12%
Water soluble	0.12%
Chelated by DTPA	0.12%
Manganese (Mn)	0.06%
Water soluble	0.06%
Chelated by EDTA	0.06%
Boron (B)	0.02%
Water soluble	0.02%
Copper (Cu)	0.015%
Water soluble	0.015%
Chelated by EDTA	0.015%
Molybdenum (Mo)	0.010%
Water soluble	0.010%
Zinc (Zn)	0.015%
Water soluble	0.015%
Chelated by EDTA	0.015%

NO
SULPHUR

**FOR ORCHIDS IN FLOWER USE CAMPBELL ORCHID "A" FERTILISER
N-P-K 11-13-16 WITH TRACE ELEMENTS.**

Nitrogen (N) present as Nitrate form	3%
Nitrogen (N) present as Ammonium form	2%
Nitrogen (N) present as Urea	25%
Total Nitrogen (N)	30%
Total Phosphorus (P) Water Soluble present as	
Mono Ammonium Phosphate	4.4%
Total Potassium (K) present as Potassium Nitrate	8.3%
Zinc (Zn) present as Zinc Sulphate	0.01%
Copper (Cu) present as Copper Sulphate	0.01%
Cobalt (Co) present as Cobalt Sulphate	0.002%
Molybdenum (Mo) present as Sodium Molybdate	0.01%
Iron (Fe) present as Iron EDTA	0.03%
Boron (B) present as Boric Acid	0.006%
Manganese (Mn) present as Manganese Sulphate	0.05%

NO Ca or Mg

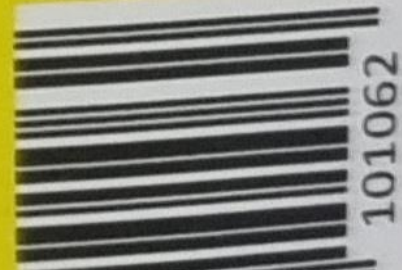
EQUAL NITRATE & AMMONIUM

AFTER FLOWERING USE CAMPBELL ORCHID "B" FERTILISER N-P-K 30-4-8 WITH TRACE ELEMENTS.

Nitrogen (N) present as Nitrate form	5.9%
Nitrogen (N) present as Ammonium form	5.9%
Total Nitrogen (N)	11.8%
Total Phosphorus (P) Water Soluble present as Mono Ammonium Phosphate	13.2%
Total Potassium (K) present as Potassium Nitrate	16.6%
Zinc (Zn) present as Zinc Sulphate	0.01%
Copper (Cu) present as Copper Sulphate	0.01%
Cobalt (Co) present as Cobalt Sulphate	0.002%
Molybdenum (Mo) present as Sodium Molybdate	0.01%
Iron (Fe) present as Iron EDTA	0.03%
Boron (B) present as Boric Acid	0.006%
Manganese (Mn) present as Manganese Sulphate	0.05%

CONDITIONS OF SALE

The buyer relies on their own skill or judgment in purchasing this product and in deciding that it may be suitable for the above mentioned purpose.



NO Ca or Mg



AMMONIUM
+
NITRATE

SULPHATE
BASED

NO Ca

SULPHATE BASED

NITROGEN(% W/W)			PHOSPHORUS (P) TOTAL(% W/W)	POTASSIUM (K)			
Nitrate-N	Urea-N	TOTAL-N		Nitrate-K	Phos.-K	Sulph.-K	TOTAL-K
3.0	12.0	15.0	2.0	9.1	4.1	11.8	25.0

TRACE ELEMENTS (% W/W)							
S	Mg	Fe	Mn	Zn	Cu	B	Mo
2.80	0.06	0.05	0.24	0.03	0.02	0.02	0.0014

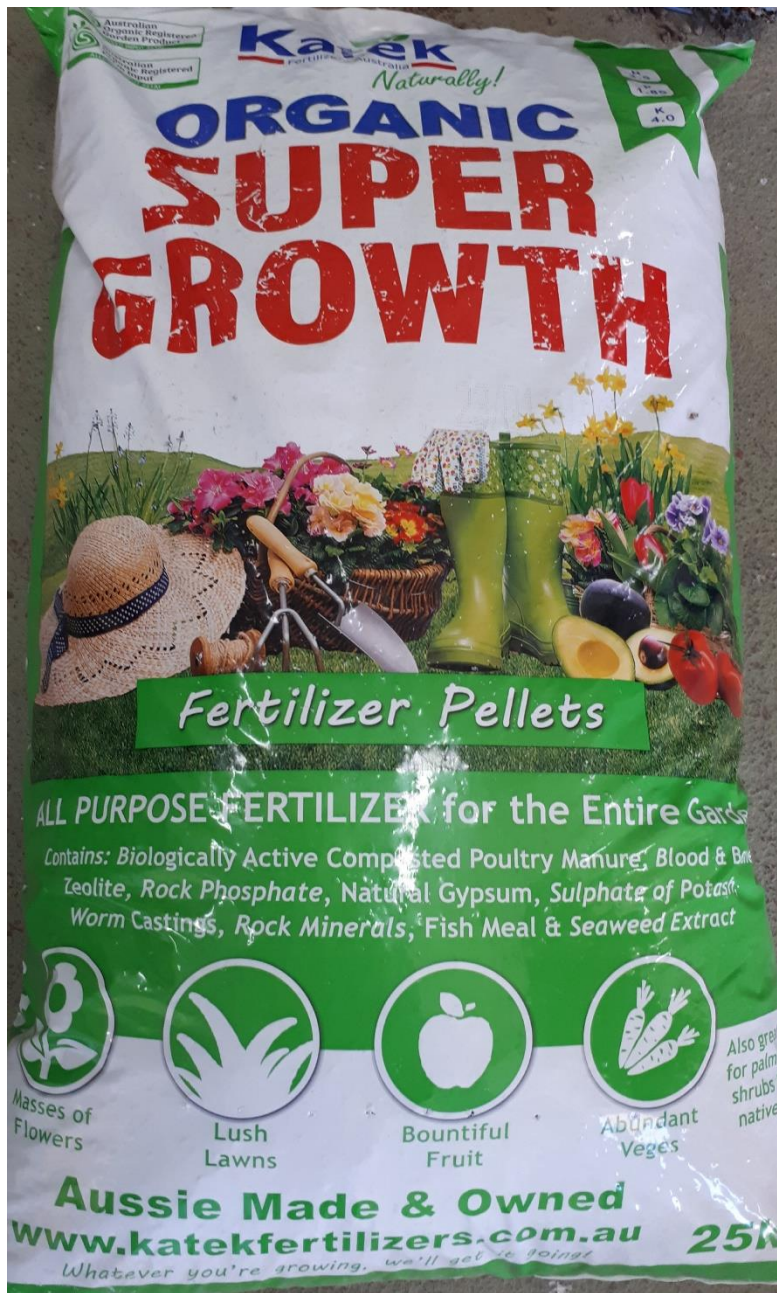
Added Vitamin B1
and Amino acid to
enhance nutrient uptake
Nitrogen to Phosphorus (N:P) ratio 7.5:1

**AMMONIUM
&
NITRATE**

NO Ca



SOME Ca & GOOD Mg





• ○

Osmocote Exact Standard 12 - 14 months

15 - 3.9 - 9.1 + 1.2Mg + TE Controlled Release Fertiliser

Controlled Release

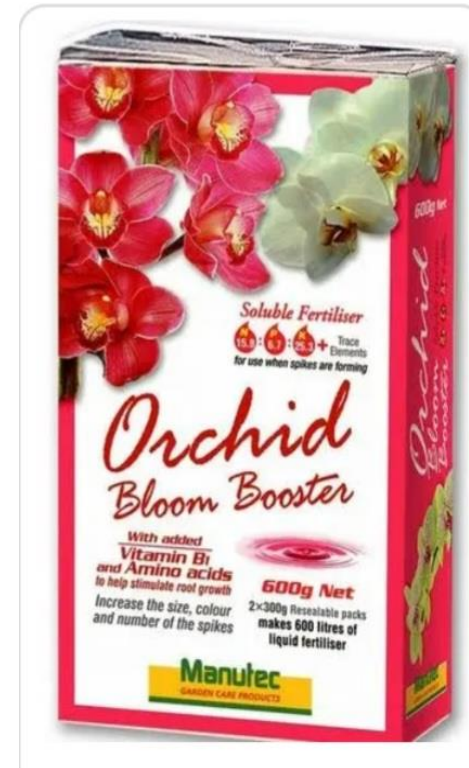


Fernland
A GARDEN SOURCE

15kg Peters Excel CalMag Finisher 13-2-17+5Ca

\$167.81 (Inclusive GST)

(Chelates)



Sulphates

NOTE: DO NOT MIX



PUMP UP SPRAYER

LIQUID FEED – ROOTS ONLY

- **MANUTEC 0.5g/L x 6**
- **AUXINEONE 0.25g/L x 3**

FOLIAR SPRAY ONLY

- **PETER'S FINISHER 0.5g/L x 6**
- **AGRI-FOS 600 0.5m/L x 8**
- **EPSOM SALTS 2g/L x 4**

TOP OF POT ONLY

- **OSMOCOTE 12-14 MNTH
SEPT/OCT or AT POTTING**



OPEN MIX - HEALTHY ROOTS FOR NUTRIENT UPTAKE



Q calcium deficiency sympt...



All

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Calcium, once deposited in plant tissue, is immobile (non-Translocatable) so there must be a constant supply for growth. Deficiency causes stunting of new growth in stems, flowers and roots. Symptoms range from **distorted new growth to black spots on leaves and fruit. Yellow leaf margins may also appear.**



<https://fameorchids.com/index.php>



[Plant Nutrition - Fame Orchids Nursery](#)



About featured snippets



Feedback

People also ask



How do you know if your orchid has a calcium deficiency?



How do I add calcium to my orchid?



Discover



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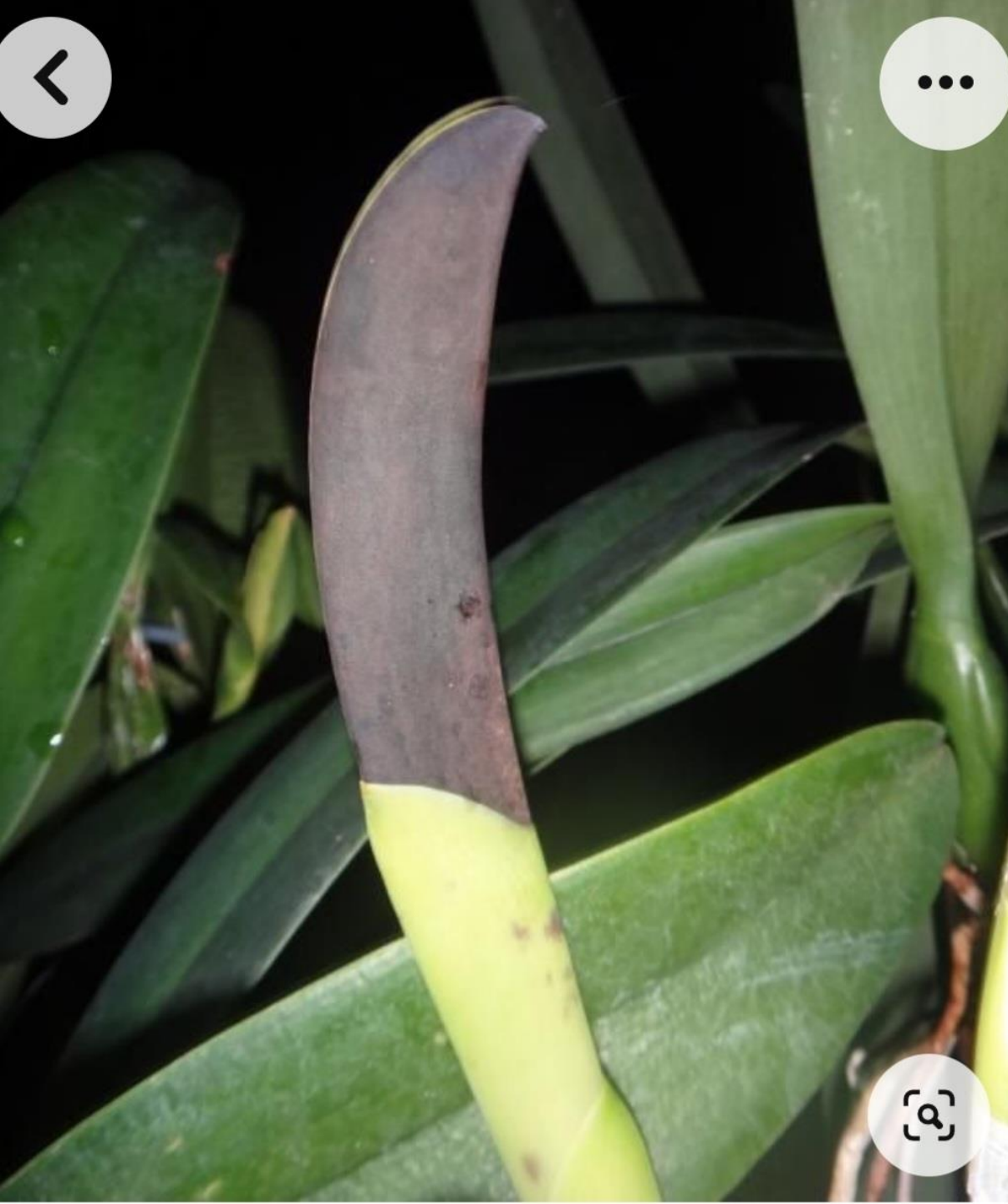
Examples of Calcium Deficiency in Cattleyas



The most rapidly expanding tissue is affected first,
Such as new growths and leaves



You might think your plant has black rot, but this
Problem is physiological rather than pathological



**ACUTE CALCIUM
DEFICIENCY**

CATTLEYA



ACUTE CALCIUM DEFICIENCY

CATTLEYA





Flower Sheath

**CALCIUM
DEFICIENCY**

CATTLEYA



LOW CALCIUM

ONCIDIUM HYBRID



LOW CALCIUM

**PHALAENOPSIS
HYBRID**



LOW CALCIUM

**PHALAENOPSIS
bellina**

Examples of Magnesium Deficiency



Cattleyas grown in bright light with insufficient magnesium may exhibit mottling indicative of chlorophyll damage



Leaves may turn a reddish purple after exposure to cold if they are magnesium deficient. Correct this with a megadose of Epsom Salts (1 tsp/gal).



LOW MAGNESIUM

PINEAPPLE



**PHALAENOPSIS LEAF
DROP OF OLD LEAVES
IN JAN/FEB**

**LESS EACH YEAR WITH
ADEQUATE CALCIUM**



EXPOSED TO NORTHERLY WINDS



Live



157/5 (41.2)



PAK chose to bat

EXPOSED TO WINDS NORTH & SOUTH



Live

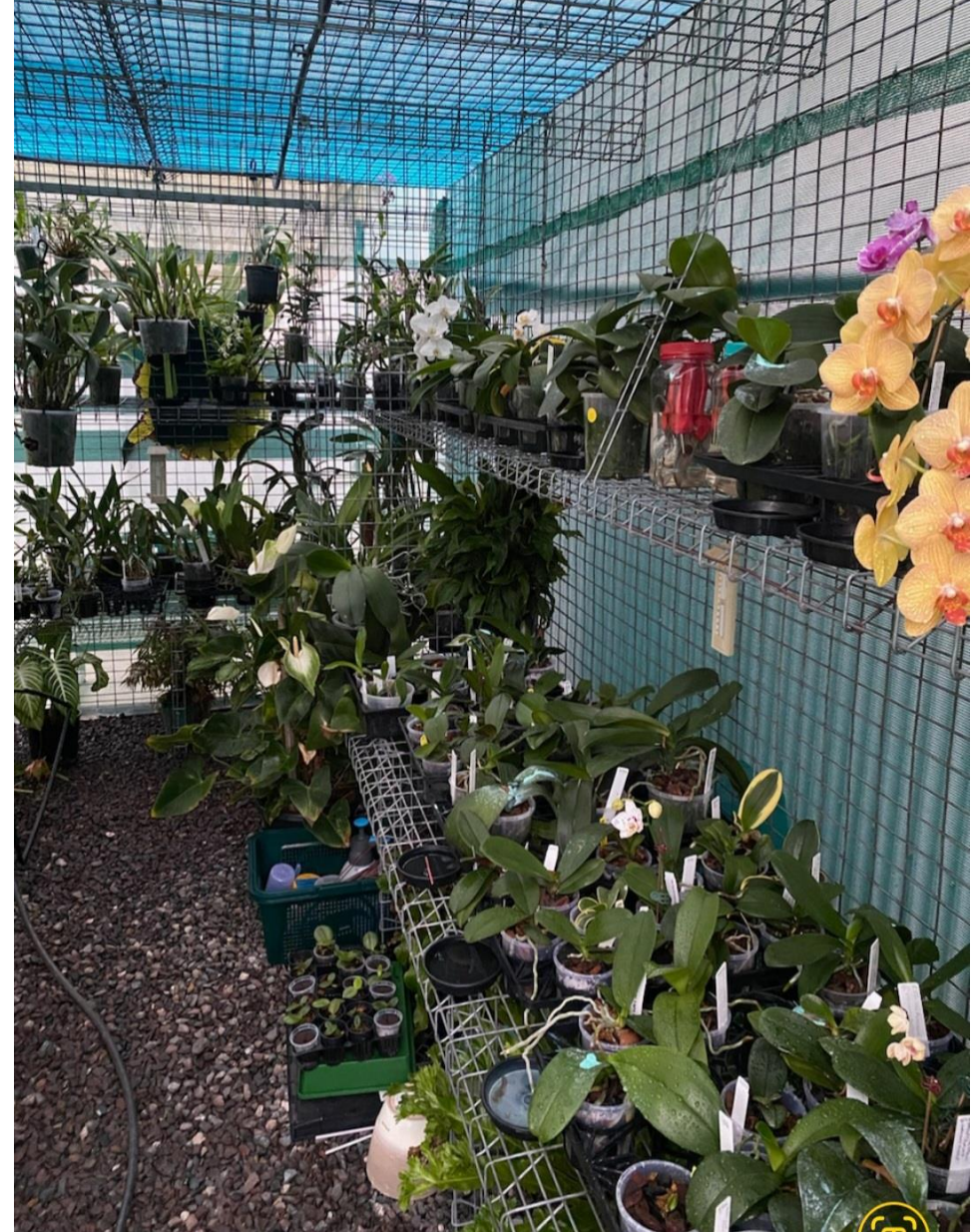
🇵🇰 157/5 (41.2)

PAK chose to bat

**OPEN ENDS – WIND TUNNEL
CAN FERTILISE OFTEN**



**PROTECTED FROM ALL WINDS
TAKE AGES TO DRY OUT
FERTILISE LESS OFTEN**



MORE FOLIAR - 'NITRATES'
LESS ROOT DRENCHES – 'AMMONIUMS'

DO NOT MIX
ANY FERTILISER CONTAINING
CALCIUM NITRATE
WITH
ANY 'SULPHATE FERTILISER'
AS IT FORMS A CALCIUM
SULPHATE (GYPSUM) SLUDGE

FERTILISER	NUTRIENT CONTENT					
	N	P	K	Ca	Mg	S
Calcium Nitrate	15.5			19		
Magnesium Nitrate	10.5				9.4	
Potassium Nitrate	13.0		45			
Magnesium Sulphate (Epsom Salts)					9.8	13
Potassium Sulphate			42			17
Calcium Sulphate (Gypsum)				23.3		18.6
Lime - Agricultural				37-40		
Dolomite Lime				22.7	11.8	
Calcium Ammonium Nitrate	27			8		
Urea	46					
Ammonium Nitrate	34					
Ammonium Sulphate	20.2					24
MAP - Mono Ammonium Phosphate	10	21.9				
DAP - Di Ammonium Phosphate	17.7	20				
Single Super Phosphate (SSP)		8.8		21.9		11

CONCLUSION

1. Always read the label and know what you are using
2. Fertilise little & often when watering is due in place of just water
3. Leach water without fertiliser every 6-8 weeks
4. Commence fertilising after flowering with the start of new growth in spring
5. Reduce high Nitrogen fertilisers in late summer/autumn to encourage flowering
6. Combination of Nitrate and Ammonium suits both tank/rain water or town water – potting mix pH!!

CONCLUSION cont.

7. **Peter's Cal Mag Finisher** ideal as foliar to supply Ca, Mg, K & Nitrates – 0.5g/L monthly winter & fortnightly summer (500g & 1kg tubs available from 'Orchid Den' & Red Fox Orchids)
8. **Neutrog Australia – 'Strike Back for Orchids Liquid'**
Ideal applied to roots to supply Ca, Mg, Sulphates and Ammonium - 2ml/L monthly winter & fortnightly summer (available from Bunnings)
9. Try soaking bark in pre-dissolved natural Gypsum 2.6g/litre to supply Calcium
10. OR consider fortnightly foliar sprays of Calcium Nitrate 0.5g/L plus Magnesium Nitrate 0.25g/L during active growth spring to late summer to supply Ca & Mg